

ton (1965). To decide between the disordered and the centered-hydrogen models for the structure of RDP, Hamilton's significance test was applied as follows: number of reflections = 129, number of variables = 25 (disordered model), number of variables = 22 (centered model); hypothesis: 'the hydrogen is centered on the O-H-O bond'.

The dimension of the hypothesis is  $25 - 22 = 3$  and the number of degrees of freedom for the refinement is  $129 - 25 = 104$ . The value of  $R$  obtained by interpolating the appropriate table for testing the  $R$ -factor ratio (Hamilton, 1965) is  $R_{3, 104, 0.005} = 1.07$ , while the observed  $R$ -factor ratio is

$$R_{\text{obs}} = \frac{R_w(\text{centered})}{R_w(\text{disordered})} = \frac{0.0498}{0.0457} = 1.09.$$

This indicates that the above hypothesis can be rejected in favor of the disordered hydrogen model at a significance level of better than 0.5% (or at a confidence interval of higher than 99.5%). In other words, the results of our experiment favor the disordered-hydrogen model for the structure of paraelectric  $\text{RbH}_2\text{PO}_4$  in which the proton is displaced on either side of the bond center at positions which are 0.41 Å apart.

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**1-Phenyl-4,5-(D-glucufurano)imidazolidine-2-thione: erratum.** By R. JIMÉNEZ-GARAY, A. LÓPEZ-CASTRO and R. MÁRQUEZ, *Departamento de Óptica y Sección de Física del Departamento de Investigaciones Físicas y Químicas de la Universidad de Sevilla, Spain*

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A new table of torsion angles for the title compound [Jiménez-Garay, López-Castro & Márquez, *Acta Cryst.* (1976), **B32**, 2115–2118] is given.

In Fig. 1 of the paper on the title compound (Jiménez-Garay, López-Castro & Márquez, 1976) atoms C(10) and C(11) are interchanged with respect to their atomic parameters (Table 1); therefore, the selected torsion angles have been recalculated and are reported in the new Table 4.

Table 4. Selected torsion angles (°)

N(1)–C(1)–C(2)–C(6)	–176.6	O(3)–C(12)–C(11)–O(1)	–115.8
C(2)–C(1)–N(1)–C(9)	111.8	C(12)–C(11)–O(1)–C(9)	163.8
C(1)–N(1)–C(9)–C(7)	–175.1	C(11)–O(1)–C(9)–C(8)	–22.7
S–C(7)–N(1)–N(2)	179.3	N(1)–C(9)–C(8)–C(10)	–119.4
O(2)–C(10)–C(8)–C(11)	–116.8	O(4)–C(13)–C(12)–C(11)	–170.1
O(1)–C(9)–C(8)–C(10)	–1.1	O(3)–C(12)–C(11)–C(13)	–177.0
C(9)–C(8)–C(10)–C(11)	22.5	C(2)–C(1)–N(1)–C(7)	–62.5
C(8)–C(10)–C(11)–O(1)	–36.3	C(1)–N(1)–C(9)–C(8)	–176.6
C(10)–C(11)–O(1)–C(9)	38.0	N(1)–C(9)–C(8)–O(1)	–118.3
C(13)–C(12)–C(11)–O(1)	61.2		

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